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Original Research Article

Relation Between Estimated Creatinine Clearance and Acid-Base Balance Status in CKD Patients- An Observational Cross Sectional Study

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Abstract

Aim: Study on Acid-Base Balance Disorders and the Relationship between Its Parameters and Creatinine Clearance in Patients with CKD.

Methods: A prospective observational study was conducted in the Department of Medicine, Darbhanga Medical College and Hospital, India, after taking the approval of the protocol review committee and institutional ethics committee. 220 patients with CKD were included in this study. CKD Patients who were not undergoing treatment or dialysis and Aged between 18-70 years were included in this study.

Results: The 220 CRF patients (male: 140, female: 80), aged 18 to 70 years were participated. The average age of patients was: 47.95 ± 14.03 . Patients aged 50-60 accounted for the highest proportion (20%). CrCl decreased in comparison with normal values and decreased with the stage of renal failure (P < 0.001). The level of creatinine, urea, uric acid increased with the stage of renal failure (P < 0.05). The mean value of protein and albumin did not change significantly (P > 0.05). Among CRF patients, metabolic acidosis is predominantly 75%, while other types of disorders account for a very low rate. This disorder gradually increases with the stage of renal failure. pH, tCO₂, HCO₃⁻ and BE had a moderate positive correlation, only PCO₂ have a weak positive correlation with creatinine clearance.

Conclusion: In the study of acid-base balance disorder, we found that different valuable biochemical indicators in CRF patients were lower than the normal range which gradually decreased based on the disease severity and stage of renal failure.

Keywords: CRF, acid-base balance, creatinine clearance

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Introduction

The kidney plays a major role in the maintenance of systemic acid-base balance, but the ability of the kidney to excrete ammonium or reabsorb bicarbonate in response to daily acid load is impaired in chronic kidney disease (CKD).¹ Therefore, metabolic acidosis, usually indicated by a low serum bicarbonate level, is common in CKD. The prevalence of metabolic acidosis depends on the definition used, for example, when metabolic acidosis was defined as a serum bicarbonate concentration of <22 mmol/L, 2.3 to 13% of patients with CKD stage 3 and 19 to 37% of patients with CKD stage 4 exhibited metabolic acidosis.² Notably, metabolic acidosis in CKD is associated with chronic inflammation, muscle protein degradation with muscle wasting, bone disease, impaired glucose tolerance, impaired albumin synthesis, accelerated CKD progression, and heart disease.³

Acid-base balance is a dynamic due to the body is an open system, so it tends to change as external factors such as food, water, or internal factors, including cellular metabolites. However, acid-base balance status is quickly recovered by the activities of the buffer sys- tem and some vital organs, especially the kidneys. To maintain acid-base balance, the kidneys excrete acid generated by metabolism and reabsorb bicarbonate. The up-take of bicarbonate from glomerular filtration is very important.⁴ Calling attention, CRF is also one of the important factors to cause the acid-base balance disorders. When disturbed, a series of extracellular pH-dependent metabolites are affected. In this line, the acid-base balance mechanism also attempts to reach to the normal pH. Although, the buffers work quickly but are limited by kidneys related slow responds, which have a longer and more effective consequent.^{5,6} However, if the main factor of the acid-base balance disorder is not eliminated, it would gradually in- crease, and cause irreparable damage for the patients.

Materials and Methods

A prospective observational study was conducted in the Department of Medicine, Darbhanga Medical College and Hospital, India, after taking the approval of the protocol review committee and institutional ethics committee. 220 patients with CKD were included in this study.

Inclusion Criteria:

- CKD Patients who were not undergoing treatment or dialysis
- Aged between 18-70 years. Exclusion criteria
- Patients with CVD
- Active cancer

Statistical Analysis

All data were analyzed by the Statistical Package for Social Science (SPSS) version 21.0 (IBM, USA). Continuous variables were described as mean and standard deviation (SD). Two or more continuous variables were evaluated by t-test or one-way ANOVA. Categorical variables were described as frequency and percentage and two or more categorical variables were evaluated by the chi-square test.

ISSN: 0975-3583, 0976-2833 VOL 12, ISSUE 04, 2021

Results

In this study, the 220 CRF patients (male: 140, female: 80), aged 18 to 70 years were participated. The average age of patients was: 47.95 ± 14.03 . Patients aged 50-60 accounted for the highest proportion (20%). The youngest patient was 20 and the oldest patient was 70 years old. To note, none of the patients had stage I CRF, and most male patients had stage IV disease (50%), while most female patients had stage IIIb (40%) and the rate of CRF in patients aged from 30 to 40 was 20%. The blood biochemical parameters have been shown in Table.2 CrCl decreased in comparison with normal values and decreased with the stage of renal failure (P < 0.001). The level of creatinine, urea, uric acid increased with the stage of renal failure (P < 0.05). The mean value of protein and albumin did not change significantly (P > 0.05).

Results of Acid-Base Balance Parameters

The results of acid-base balance parameters have been shown in table 3. Among CRF patients, metabolic acidosis is predominantly 75%, while other types of disorders account for a very low rate. This disorder gradually increases with the stage of renal failure. Table 4

Relationship between Acid-Base Balance Parameters and Creatinine Clearance pH, tCO_2 , HCO_3^- and BE had a moderate positive correlation, only PCO_2 have a weak positive correlation with creatinine clearance. Table 5

Table 1: Wethodology and blood Test Normal Values						
Parameters	Units	Normal Value	Methodology			
Urea	mg/dl	2.5 - 7.5	Urease			
Creatinine men	mg/dl	90 - 120	Jaffe			
Creatinine women	mg/dl	80 - 110	Jaffe			
Creatinine clearance men	mL/min	97 - 137	Calculated			
Creatinine clearance women	mL/min	88 - 128	Calculated			
Protein	g/L	60 - 80	Biuret			
Albumin	g/L	35 - 50	BCG			
Uric acid men	µmol/L	180 - 420	Uricase			
Uric acid women	µmol/L	120 - 360	Uricase			
рН		7.38 - 7.42	ISE			
PCO2	mmHg	35 - 40	ISE			
-	mmol/L	22 - 26	ISE			
tCO2	mmol/L	25 - 30	ISE			
BE	mmol/L	0 ± 2	Calculated			

Table 1: Methodology and Blood Test Normal Values

Table 2: Biochemical Parameters in CRF Patients^{a,b}

Parameters	GD II (1)	GD IIIa (2)	GD IIIb (3)	GD IV (4)	
	(N = 42)	(N = 38)	(N = 65)	(N = 75)	
Urea (mg/dl)	$15.57 \pm 6.20^{\circ}$	$25.75\pm8.85^{\rm C}$	$31.86 \pm 13.58^{\circ}$	$42.83 \pm 10.67^{\circ}$	
Creatinine	216.00 ± 45.21	391.0 ± 62.27	666.82 ± 11.91	1221.3 ± 412.4	
(mg/dl)					
CrCl (mL/min)	$32.13 \pm 4.46^{\circ}$	$15.67 \pm 1.75^{\circ}$	$7.17 \pm 1.18^{\circ}$	$2.81 \pm 1.10^{\circ}$	
Uric acid	486.95 ± 105.12	571.35 ±	587.28 ± 142.63	581.97 ± 138.50	
(µmol/L)		129.70	367.26 ± 142.03	301.97 ± 130.30	
Proteine (g/L)	$63.78\pm8.28^{\mathrm{B}}$	64.18 ± 7.95^{B}	$64.68\pm8.55^{\mathrm{B}}$	$63.25\pm6.39^{\mathrm{B}}$	

Albumine (g/L	L) 32.94 ± 6.16		33.54 ± 4.45		32.91 ± 5.81		33.84 ± 5.29		
	Table 3: Results of Acid-Base Balance Parameters								
Parameters) II (1) = 42)	GD] (N =	IIIa (2) 38)	$\begin{array}{c} \text{GD III} \\ \text{(N = 65)} \end{array}$	• • •	GD IV (4) (N = 75)		Total (N = 220)
рН	7.3	9 ± 0.09	7.38 =	± 0.07	7.34 ± 0	0.08^{A}	7.22 ± 0.73	А	$7.37\pm0.09^{\rm A}$
PCO2 (mmHg)	36.4	47 ± 6.62	36.15	5 ± 5.86	33.48 ±	7.27	32.20 ± 7.1	0	34.27 ± 6.92
StHCO3 - (mmol/L)	24.4	47 ± 4.45	23.11	± 4.94	19.38 4.53B	±	18.35 4.84C	±	21.27 ± 5.
tCO2 (mmol/L	25.	12± 5.64	23.48	3 ± 6.06	22.11 15.97	±	18.80 ± 5.8	8	21.37 ± 6.48
BE (mmol/L)	-0.5	51 ± 5.59	-2.25	$\pm 6.19^{A}$	$-7.09 \pm$	5.90 ^B	-7.18 ± 6.32	2 ^C	-4.75 ± 6.64

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 Table 4: Types of acid-base balance disorders

Types of acid-base balance disorders	Ν	%
Normal	15	6.82
Metabolic acidosis	165	75
Respiratory acidosis	3	1.36
Metabolic alkalosis	2	0.91
Respiratory alkalosis	1	0.45
Metabolic acidosis + respiratory alkalosis	14	6.36
Metabolic acidosis + respiratory alkalosis	14	6.36
Metabolic alkalosis + respiratory alkalosis	3	1.36
Metabolic alkalosis + respiratory acidosis	3	1.36
Total	220	100

Parameters		r	Р	The Regression Equation
pН	Ccr	0.370	0.0001	Y = 40.574x - 286.323
PCO2 (mmHg)	Ccr	0.278	0.003	Y = 0.381x - 0.408
tCO2 (mmHg)	Ccr	0.367	0.0001	Y = 0.643x - 1.013
StHCO3 -	Ccr	0.394	0.0001	Y = 0.913x - 6.567
(mmol/L)				
BE (mmol/L)	Ccr	0.403	0.0001	Y = 0.725x + 16.453

Abbreviations: BE, base excess; CrCl, creatinine clearance; PCO2, pressure of carbon dioxide; r, correlation coefficient; StHCO3, standard bicarbonate; tCO2, total carbon dioxide.

Discussion

In this study, the 220 CRF patients (male: 140, female: 80), aged 18 to 70 years were participated. The average age of patients was: 47.95 ± 14.03 . Patients aged 50-60 accounted for the highest proportion (20%). The youngest patient was 20 and the oldest patient was 70 years old. However, it is more likely to occur in the elderly population, due to the co-morbidity of

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other diseases such as diabetes and hypertension.^{7,8} In general, it seems that the cost of treatment for patients with renal insufficiency needs more attention. However, in developing countries, health care services for people who are at risk of kidney failure is still limited. It has been previously found that urea and creatinine levels in blood and urine are important indicators to evaluate kidney function.9 According to our results, blood urea and creatinine levels are closely related to the stage of CRF progression. Overall, the kidney has the functional role in maintaining body homeostasis, therefore when the kidney failure occurs, the metabolites level especially urea, creatinine cannot completely excreted, resulting in the accumulating toxic substances in the body and affecting the function of multiple organs which finally present with a variety of clinical symptoms. Many authors have used creatinine clearance as a standard indicator to determine the different CRF stages.^{5,10,11} It is not worthy that the blood uric acid level increased concomitantly with the CRF stage. Uric acid is a degradation product of the base purine primarily from nucleic acid. Uric acid is also excreted by the kidneys in the urine. Therefore, during renal failure, uric acid is not eliminated and accumulates in the blood. However, hyperuricemia is not only due to impaired kidney function also to other conditions such as gout, leukemia, high meat diet, alcohol consumption, etc. In our study, the blood proteins such as albumin levels in CRF patients decreased compared to normal subjects. Presumably, it could be related to the damaged glomerular membrane, which causes proteins passing through the glomerular filtration membrane into the urine. However, both blood protein and albumin levels slightly decreased with the stage of renal failure.(table 2) Acid-Base Balance Disorders in Patients with CRF

According to our findings, blood pH in CRF patients gradually decreased with the stage of renal failure. The kidneys maintain the normal state of pH concentration by bicarbonate re absorption and normalizing the acid excretion. Given that the normal urine pH is approximately 5.6 while the blood pH is between 7.38 - 7.42, it is proving that the role of the kidney in maintaining acid- base balance is undeniable.⁵ There are four types of acid-based disorders: respiratory alkalosis, respiratory acidosis, metabolic alkalosis and metabolic acidosis. When these conditions occur, to maintain normal levels of pH, the body can create an opposite counterbalance as a compensatory mechanism. For instance, if the patient has metabolic acidosis, the body will induce a respiratory alkalosis, but it rarely makes our pH return to normal at 7.4.9 To maintain the acid-base balance, the activity of the buffers and lungs is fast but short and limited. In this regard, although the kidney function is slow but most effective and long-lasting.^{11,12} as shown in table 3, the amount of PCO₂ decreased with the stage of renal failure. In fact, CO₂ is excreted through the respiratory system. Therefore, the compensatory mechanism induced by bicarbonate buffer and the lungs reduce both the stability of the blood acidity and limitation in the pH reduction due to kidney failure.⁹ However, the complementary activity of the buffer system and the lungs is limited, so if the patient has severe renal dysfunction, the pH is greatly reduced, and subsequently, the clearing operation would not be repeated as usual, and the pH decreases gradually according to the severity of chronic renal failure.¹⁰ On the other hand, the PCO₂ level determines respiratory activity. Given that the reversible reaction between CO₂ and H₂O, which produces H₂CO₃, the respiratory system can adjust the pH level indirectly. For example, once the pH de- creased, the respiratory system releases more CO₂ by increasing breathing frequency, thereby reducing the acidity in the body, leads the pH rising as well as the PCO₂ level reduction.¹² Under the normal circumstance, the kidney excretes the acid (H^+) to regenerate and reabsorb HCO_3^- , during the kidney failure, both acid excretion and re-absorption of HCO_3^- are reduced. In our study, actual HCO_3^- and tCO_2 also decreased.¹³ On the other hand, tCO₂ is a value to represent the total amount of CO₂ in the blood, including dissolved CO₂, carbonic acid (H₂CO₃), and HCO₃⁻. tCO₂ is also a valuable

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marker in distinguishing between acid-base disorders due to some metabolic or respiratory reasons.¹³ BE is excess or deficiency of acid or base in the blood, considered as an important indicator of acid-base balance disorders, which is also another value of the difference between total base buffer (BB) and BB. In our study, BE in patients with CRF is -4.75 ± 6.64 mmol/L. In stage II renal failure, it is -0.51 ± 5.59 mmol/L, but in stage IV renal failure, BE is $-7.18 \pm$ 6.32^C mmol/L, lack of base and excess acid, need to be timely treated. Based on pH, PCO₂, and HCO₃⁻, most patients with CRF have metabolic acidosis. According to Kopple et al. reports, the pH of cells and other physiological fluids is an important biological constant of the body, which fluctuates in a very narrow range to sustain life.¹⁰ Effros and Widell also suggested that in patients with CRF when the glomerular filtration rate decreased < 30 mL/min, blood urea increased> 14.3 mmol/L and creatinine increased > 354 μ mol/L, the renal function decreased, acid is not eliminated and the ability of bicarbonate re-absorption is reduced leading to metabolic acidosis.^{14,15} Using the Davenport diagram we also evaluated the state of acid-base balance disorder in the area of metabolic acidosis (data not shown). The indicators of acid-base balance disorders of renal failure different stages gradually change and pH, PCO₂ and HCO de- crease while the level of metabolic acidosis increases. Relationship between Acid-Base **Balance Parameters and Creatinine Clearance**

Many studies have shown a relationship between the various parameters of acid-base balance and creatinine clearance.¹² In fact, acid-base balance parameters have a positive correlation with creatinine clearance, (correlation coefficient $r = 0.367 \quad 0.403$). Among them, only PCO₂ is less correlated with creatinine clearance (r = 0.278, P < 0.01). The creatinine clearance not only helps clinicians to understand the level of acid-base balance disorders, but also has ability to predict the stage of kidney failure.^{5,8} Ultimately, by monitoring the blood creatinine, we can manage the acid-base balance disorders to improve the quality of life for patients.

Conclusion

In the study of acid-base balance disorder, we found that different valuable biochemical indicators in CRF patients were lower than the normal range which gradually decreased based on the disease severity and stage of renal failure. Most patients with CKD have metabolic acidosis, and the mentioned parameters are positively correlated with creatinine clearance coefficient.

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