Original Research Article

Materials And Methods of Limnological Study of Ganga River Water Within District Haridwar

Aditi^{1*}, Dr. Ravi Bhatnagar²

^{1*}Research Scholar, Department of Zoology, School of Science, Sunrise University, Alwar, Rajasthan, India Email: ^{1*}<u>04aditiyadav@gmail.com</u>
²Department of Zoology, School of Science, Sunrise University, Alwar, Rajasthan, India Email: ²srualwar6560@gmail.com

Corresponding Author: Aditi^{1, 1*}04aditiyaday@gmail.com

ABSTRACT

Haridwar is situated on the bank of river Ganga at the foothills of the Shivalik range of mountains that constitute the outer Himalayas. Haridwar city lies at an elevation of 965ft from the sea level and between latitude 20⁰, 58' N and Longitude 78⁰, 13' E. According to Hindu mythology, Hardwar is one of the four sites where drops of the elixir of immortality (AMRITA) accidentally spilt over from the pitcher in which it was being carried (KUMBHA) away by the celestial bird Garuda, after the Samudra Manthan (charging of sea). These four places are Haridwar, Allahabad, Ujjain and Nasik where the famous **"Kumbh Mela"** is held. It is believed that taking a bath here purifies the soul and opens the way for the ultimate freedom, Moksha.

Keywords: Physico-chemical parameters, Ganga River, Haridwar, Sapta Rishi Ghat, Har ki Pauri.

1. INTRODUCTION

Hydrobiology is the science of life and life processes in water. Water is not only a major component of the environment but also the best solvent and a medium in which all organisms depend for their existence. A fresh water body, which fulfils a variety of human needs is full of value only when it is not abused and polluted. Hydrobiology deals with the details of various forms of aquatic life such as algae phytoplankton, periphyton, lithophytes and benthos, zooplankton, fishes and other groups of living organisms. Phytoplankton, periphyton and benthic algae communities represent the major producers in aquatic ecosystems and Diatoms are good indicators of water quality as pointed out by Odum (1971). Hundling (1971) has described the algae as an important producer component of the littoral zone of water bodies. The freshwater limnology plays an important role in the decision-making process for problems like dam construction, pollution control and aquaculture practices. The river catchment from the mountains to the sea is a single ecosystem by itself, linked to other catchment ecosystems through terrestrial corridors, atmospheric corridors and subterranean

corridors. Fresh water has been of vital importance to man and animals for the sustenance of life and maintaining the balance of nature. Freshwater constitutes only about three per cent of the total water present on the earth.

Study Area

Haridwar is situated on the bank of river Ganga at the foothills of the Shivalik range of mountains that constitute the outer Himalayas. Haridwar city lies at an elevation of 965ft from the sea level and between latitude 20⁰, 58' N and Longitude 78⁰, 13' E. According to Hindu mythology, Hardwar is one of the four sites where drops of the elixir of immortality (AMRITA) accidentally spilt over from the pitcher in which it was being carried (KUMBHA) away by the celestial bird Garuda, after the Samudra Manthan (charging of sea). These four places are Haridwar, Allahabad, Ujjain and Nasik where the famous **"Kumbh Mela"** is held. It is believed that taking a bath here purifies the soul and opens the way for the ultimate freedom, Moksha.

Sampling Stations

The following seven sites were selected to evaluate the impact of pilgrims' activities on the water quality of the River Ganga in Haridwar city.

- 1. Sapta Rishi Ashram
- 2. Har-Ki Pauri
- 3. Prem Nagar Ashram Ghat
- 4. Pul Jatwada

1. Sapta Rishi Ashram: Sapta Rishi Ashram ghat is an ancient ashram famous for having hosted seven (sapt) sages (rishis), namely, Kashyapa, Vashisht, Atri, Vishwamitra, Jamadagi, Bharadwaja and Gautam. To avoid any disturbance to the seven sages meditating in the ashram, it is believed that the river Ganga split itself into seven currents at this spot. Hence, this ghat is also referred to as Sapt Sarovar or seven streams. The Sapt Rishi Ashram is located about 5 km away from Har-ki-Pauri. This site is exempt from heavy visitor load and has been treated as an entry point to Haridwar city for river Ganga and selected as a reference site.

2. Har- Ki- Pauri: This Ghat/platform is the most popular bathing site and it is the busiest ghat/platform in the city. It is situated on the right bank of the Upper Ganga canal about 2.5 km. away from Haridwar railway station. This spot is famous for its religious inviolability and as a bathing ghat cum tourist place. During the festive days, a congregation of pilgrims occurs and offers different types of offerings to the river Ganga and takes holy dips throughout the length (Approximately 750m) and breadth (100m) of bathing ghats (platforms) at Har Ki Pauri.

3. Prem Nagar Ashram Ghat: Shri Prem Nagar Ashram is the premier Ashram established by Shri Hans Ji Maharaj and further developed by Jagat Janani Shri Mata Ji and Shri Satpal Ji Maharaj and It is situated on the banks of the Ganga canal. The twin peaks of Chandi Devi and Mansa Devi, capped by temples, are visible from the Ashram. The sounds of nearby temple bells and devotional songs are carried on the morning and

evening breezes, mingling with birdsong and augmenting the peaceful and contemplative atmosphere of the Ashram.

4. Pul Jatwada: This ghat is situated in the Jwalapur town of Haridwar. This ghat/platform is mostly used by local people for bathing. Two domestic sewer drains are continuously falling near this ghat/ platform.



Fig:-1: Map of Haridwar city showing location the of study site

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Plate 1. Bathing ghat of Sapta Rishi Ashram (Study Site- I)



Plate 2. Bathing ghat of Har Ki Pauri (Study Site- II)



Plate 3. Bathing ghat of Prem Nagar Ashram (Study Site- III)



Plate 4. Bathing ghat of Pul Jatwada (Study Site- IV)

Sampling Schedule

The monitoring of water quality was conducted every month during the years 2016 and 2017. While Planktonic biomass, Planktonic diversity and Ichthyofaunal diversity were analyzed seasonally during the study period

Parameters Studied

Water quality parameter: The study was carried out by systematic collection of water samples from four spots namely Sapta-Rishi Ashram Ghat, Har-Ki-Pauri, Prem Nagar Ashram Ghat and Pul Jatwada. The samples were collected monthly for the following parameters

Physico-chemical Parameters

- 1. Temperature (°C)
- 2. Velocity (m/sec)
- 3. Total Solids (T.S.) mg/l
- 4. Total Dissolved Solids (T.D.S.) mg/l
- 5. Total Suspended Solids (T.S.S.) mg/l
- 6. Turbidity (NTU)
- 7. Transparency (cm)
- 8. pH
- 9. Dissolved Oxygen (DO) mg/l
- 10. Biological Oxygen Demand (BOD) mg/l
- 11. Total Hardness (mg/l)
- 12. Chlorides (mg/l)
- 13. Alkalinity (mg/l)

Biological Parameters: The following biological parameters were analyzed during the both years 2016 and 2017.

- 1. Phytoplankton Diversity
- 2. Zooplankton Diversity
- 3. Fish Fauna

2. SAMPLING AND ANALYTICAL METHODS

The following physicochemical parameters were analyzed by the standard methods of **APHA (1995) and Trivedi & Goel (1986).**

Water Temperature

Temperature is important for its effect on the physical, chemical and biological processes. The temperature was analyzed by using a water and soil analysis kit (model 191 E).

Surface water: For the determination of surface water temperature, a water sample was collected in a suitable container. Soon after the collection of the sample, a mercury thermometer was inserted to take readings. Thermometers should be of small thermal capacity to attain equilibrium rapidly and must be graduated up to an accuracy of 0.1 to 0 the accuracy of the thermometer may vary depending upon the type of analysis.

VELOCITY

The surface flow method (cork method) was used to find the velocity of the Ganga River at each study site. This is a simple approach. A float (any piece of wood, plastic, etc) is thrown on the water surface. The time required for a float to travel (t), a known distance (d) is observed and the average velocity is obtained by



The factor 1.2 accounts for the fact that surface velocities are normally about 1.2 times higher. If the cross-sectional area (A) is measured, the discharge Q is given by

Q= VA TOTAL SOLIDS

Principle

Total solids as determined as the residue left after evaporation of the unfiltered sample.

Calculation Total solids mg/l = $\underline{A-B \times 1000 \times 1000}$

Where, A= Final weight of the dish in gm
 B= Initial weight of the dish in gm
 V= Volume of the sample taken in ml
TOTAL DISSOLVED SOLIDS
Principle

Total Dissolved solids are determined as the residue left after evaporation of the filtered sample.

Calculation

Total solids $mg/l = A-B \ge 1000 \ge 1000$

V

Where, A= Final weight of the dish in gm B= Initial weight of the dish in gm V= Volume of sample taken in ml

TOTAL SUSPENDED SOLIDS (TSS)

Total Suspended Solids are the difference between the total solids and total dissolved solids.

TSS mg/l =TS-TDS TURBIDITY Principle

When light is passed through a sample having suspended turbidity, some of the light is scattered by particles. The scattering of the light is generally proportional to the turbidity. The turbidity of the sample is thus measured from the amount of light scattered by the sample taking a reference with standard turbidity suspension. The determination of turbidity is interfered with by the presence of debris and other rapidly settable matter, the true colour in the sample reduces the values of turbidity. The turbidity of the Ganga water sample was determined using a digital "water and soil analysis kit" (model 191 E).

TRANSPARENCY

Principle

Light penetration in the River Ganga at each site was obtained by immersing the Secchi disc and observing its visibility. (Secchi disc is a circular disc of metal 20cm in diameter, painted matt white. Sometimes discs painted alternately black and white in radial fashion are also used. It has got a weight at the lower face to avoid a drift during lowering in water. A string is attached to it for lowering, which may be marked in centimetres.)

Calculation

Secchi disc light penetration = $\underline{A}+\underline{B}$

2

Where, A= Depth at which secchi disc disappears

B= Depth at which secchi disc reappears

pН

pH is negative \log_{10} of Hydrogen ion concentration in a solution. The digital water and soil analysis kit (model 191 E) was used to obtain the pH value of the Ganga water sample.

DISSOLVED OXYGEN (DO)

Principle

The manganous sulphate reacts with alkali (KOH or NaOH) to form a white precipitate of manganous hydroxide which in the presence of oxygen, gets oxidized to a brown colour compound. In the strong acid medium manganic ions are reduced by iodide ions which get converted into iodine equivalent to the original concentration of oxygen in the sample. The iodine can be titrated against thiosulphate using starch as an indicator. The DO was measured using the titration method.

Calculation

When the whole content has been titrated:



Where, V_1 = volume of sample bottle after placing the stopper V_2 = volume of the part of contents titrated V= volume of MnSO₄ and KI added

BIOCHEMICAL OXYGEN DEMAND (BOD)

Principle

Biochemical Oxygen Demand (BOD) is the measure of the degradable organic material present in a water sample and can be defined as the amount of oxygen required by the microorganism to stabilize the biologically degradable organic matter under aerobic conditions. The principle of the method involves, measuring the difference of the oxygen concentration between the sample and after incubating it for 5 days at 20° C.

Calculation

BOD, $mg/l = (D_0-D_5) X$ dilution factor Where, D_0 = Initial DO in the sample D_5 = DO after 5 days.

TOTAL HARDNESS

Principle

Hardness is generally caused by calcium and magnesium ions present in the water. Polyvalent ions of some other metals like strontium, iron, aluminium, zinc manganese etc. are also capable of precipitating the soap and thus contributing to the hardness. However, the concentration of these ions is very low in natural waters, therefore hardness is generally measured as concentrations of only calcium and magnesium (as calcium carbonate), which are far higher in quantities over other hardness-producing ions.

Calcium and magnesium form a complex of wine red colour with Eriochrome black-T at a pH of 10 ± 0.1 . the EDTA has a stronger affinity towards Ca⁺⁺ and Mg⁺⁺ and, therefore, by addition of EDTA, the former complex is broken down and a new complex of blue colour is formed.

Calculation

Hardness as mg/l CaCO₂ = ml EDTA used×1000 ml of sample

CHLORIDES

Principle

Silver nitrate reacts with chloride to form a very slightly soluble white precipitate of AgCl. At the endpoint when all the chlorides get precipitated free silver ions react with chromate to form silver chromate of reddish brown colour.

Procedure:

- 1. Take 50 ml of sample in a conical flask and add 2 ml of K_2Cro_4 solution.
- 2. Titrated the contents against 0.02 N AgNO₃ until a persistent red tinge appears.

Calculation:

Chloride mg/l = $\frac{(ml \times N) \text{ of } AgNO_3 \times 1000 \times 35.5}{ml \text{ of sample}}$

ALKALINITY:

Principle:

Total alkalinity is the measure of the capacity of the water to neutralize a strong acid. The alkalinity in the water is generally imported by the salts of carbonates, bicarbonates, phosphates, Nitrates, borates, silicates, etc. together with the hydroxyl Ions in the free state. However, most of the waters are rich in carbonates and bicarbonates with little concentration of other alkalinity imparting Ions.

Total alkalinity, carbonates and bicarbonates can be estimated by Titrating the sample with a strong acid (HCL or H2SO4), first to pH 8.3 using phenolphthalein as an indicator and then further to pH between 4.2 and 5.4 with methyl Orange or mixed indicator.

In the first case, the value is called phenolphthalein alkalinity (pa) and in the second case, it is total alkalinity. value of carbonates, bicarbonates and hydroxyl ions can be computed from these two types of alkalinity.

Reagents : Hydrochloric acid Methyl orange indicator

Phenolphthalein indicator Sodium carbonate.

Procedure :

1. Take 100ml of Sample in a conical flask and add 2 drops of Phenopthalien indicator.

2. If the solution remains colourless, PA=0 and total alkalinity is determined.

3. Now add 2-3 drops of methyl orange and continue the titration further until the yellow colour changes to pink at the endpoint. This is total alkalanity (TA).

Calculation:

TA as CaCo3, mg/L = $(A \times normality)$ of HCL x 1000x50 ml of sample Where A = ml of total HCL used with phenolphthalein and methyl orange TA = Total alkalanity.

Biological characteristics: The physico-chemical parameters mentioned above are influenced by the biological interaction taking place in the water body. Information on the type of aquatic organisms such as plankton, fishes etc. are perquisite in the appraisal of water quality standards. However, the difficulties involved in describing biotic communities in running water are well known; they are often marked as heterogenetic and show rapid temporal changes in the study. The following methods were considered for the determination of biotic communities.

PHYTOPLANKTON COLLECTION AND ENUMERATION

The samples of phytoplankton and zooplankton were collected monthly in the Ganga River at seven different sites by filtering 50 ltr. of subsurface water sample through plankton net made up of bolting silk cloth no. 20 (mesh size 0.06). the filtrate thus obtained was brought to the laboratories of Zoology & Environmental Sciences Gurukula Kangri University Haridwar and centrifuged at a moderately high speed, preserved in 4% formalin and ligule's solution separately for further study i.e. for qualitative and quantitative analysis. Identification was done following, Ward and Whipple (1959), and APHA-AWWA (1985).

ICHTHYOFAUNA

Fishes were identified from different strata and mainly the collection was made with the help of local fishermen and villagers using simple cloth. Fishes soon after collection were brought to the laboratories of Zoology & Environmental Sciences Gurukula Kangri University Haridwar for their proper identification following the description given by Mishra (1952), Day (1878), Jayaram (1981), Qureshi and Qureshi (1983), Srivastava (1984, 1997), Talwar and Jhingran (1991), Jhingran (1992) and Badola (2009).

3. CONCLUSION

This study is focused on the plankton diversity of the Ganga River during the year 2016. Hydrobiology is the science of life and life processes in water. Water is not only a major component of the environment but also the best solvent and a medium in which all organisms depend for their existence. A fresh water body, which fulfils a variety of human needs is full of value only when it is not abused and polluted. Hydrobiology deals with the details of various forms of aquatic life such as algae phytoplankton, periphyton, lithophytes and benthos, zooplankton, fishes and other groups of living organisms.

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